OCCURRENCE OF X-GLUTAMYL-3-CYANOALANINE

Bruno Tschiersch

Pharmakognostisches Institut der Universität Halle; Institut für Biochemie der Pflanze der DAW zu Berlin

Halle (Saale), DDR

(Received 10 February 1964)

A study of free amino acids and related compounds in seedlings of <u>Vicia sativa L</u>, and of several other plants in their relation to H¹⁴CN assimilation revealed the presence of a strongly labelled ninhydrin-reacting substance of unknown constitution, designed X₁ in preceding experiments¹,²,³. This compound is present in higher concentration only in <u>Vicia sativa</u> and near related species. In several other plants examinated until now it is detectable only after feeding H¹⁴CN to the plants. This unknown compound could be separated from glutamic acid on paper using phenol-water as solvent or by ionophoresis in phosphate buffer pH 7,2. It gave a purple colour with ninhydrin, and behaved as an α-monoamino carboxylic acid when treated successively with cupric nitrate and ninhydrin.

After extracting with 70% ethanol from finely ground seeds of Vicia sativa and partially purifying by ion exchange chromatography⁴, the acidic fraction, containing aspartic acid, glutamic acid, β -cyanoalanine, and the compound X_4 , was separated by ionophoresis. The crude material of X_4 , concentrated at room temperature under reduced pressure, was recrystallized repeatedly from the azeotrop system propan-1-ol-water, m.p. 178°. A sample of 2,0 mg was hydrolyzed in 6 N hydrochloric acid for 24 hours at 105°, and the hydrolysate was used for analysis by paper chromatography and by iono-

phoresis. Only aspartic acid and glutamic acid were found, in a ratio 1:1. Another sample was converted into the DNP-derivative, according the method of Lewy 5 . Free aspartic acid and DNP-glutamic acid were observed after acid hydrolysis. By refluxing 2,0 mg of $\rm X_1$ with 10 ml 0,1 N NaOH for 3 hours, the compound was converted into a neutral amino acid, showing a high $\rm R_1$ value on paper chromatograms run in phenol-water and in propan-1-ol-water. The DNP-derivative of this compound, isolated by paper chromatography, also yields aspartic acid and DNP-glutamic acid after hydrolysis in 6 N hydrochloric acid. But among the degradation products dinitroaniline was detectable, too. Another sample of $\rm X_1$ was treated with Sn and HCl. The reaction products were purified by ion exchange chromatography, hydrolyzed in 6 N HCl and analyzed. On the chromatograms and after ionophoretic separation only three spots were observed, corresponding with glutamic acid, aspartic acid, and $\rm \alpha, g$ -diaminobutyric acid.

The results of the experiments presented here, led us to suggest the following structure for the compound X_4 :

HOOC -
$$CH$$
 - CH_2 - CH_2 - CO - NH - CH - CH_2 - CN
 NH_2
 $COOH$

y-Glutamyl- β -cyanoalanine

These results are in agreement with those of Ressler et al. 6, who recently isolates the same compound, searching for the neurotoxic

factors of Vicia sativa.

In this connection it is remarkable that β -cyanoalanine and γ -glutamyl- β -cyanoalanine are the first detectable $H^{14}CN$ assimilation products not only in <u>Vicia sativa</u>, but also in all the other plants examinated till now^{2,3,7}. It is not yet known, if the biosynthetic pathways, observed after $H^{14}CN$ feeding to the plants, are occurring in the normal metabolism of these plants, too. Neverthe-

less it has been shown that the radioactive carbon atom of supplied hydrocyanic acid flows via β -cyanoalanine into the metabolism. In <u>Vicia sativa</u> and several other <u>Vicia</u> species β -cyanoalanine as well as γ -glutamyl- β -cyanoalanine were accumulated. Many other plants fed with H^{14} CN accumulate asparagine, synthesized from previously formed β -cyanoalanine, and in H^{14} CN fed <u>Lathyrus odoratus</u> seedlings the Lathyrus factor is one of the especially labelled compounds. Therefore it is possible that this γ -glutamyl- β -aminopropionitrile is synthesized from β -cyanoalanine via γ -glutamyl- β -cyanoalanine by a simple decarboxylation.

The author thanks Professor K. Mothes for his interest in this work.

REFERENCES

- 1 B. Tschiersch, Flora, Jena 153, 115 (1963)
- 2 B. Tschiersch, Flora, Jena 155, in press
- 3 B. Tschiersch, Phytochemistry 4, in press
- 4 P.H. Plaisted, Contrib. Boyce Thompson Inst. 19, 231 (1958)
- 5 A.L. Lewy, Nature, London 174, 126 (1954)
- 6 Ch. Ressler, S.N. Nigam, Y.H. Giza, and J. Nelson J.Amer.Chem.

 85, 3311 (1963)
- 7 B. Tschiersch, Pharmazie 19, in press